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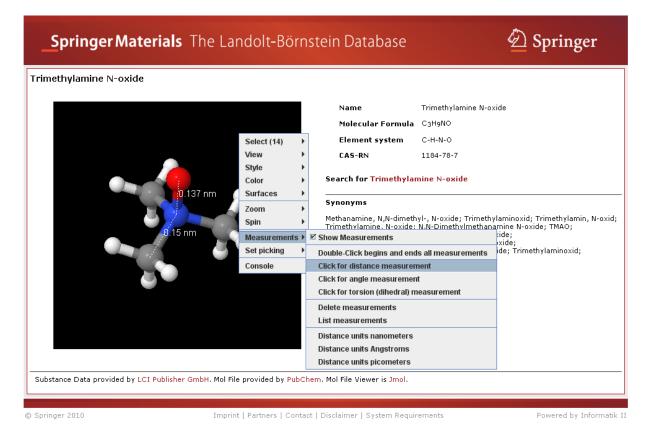
# **SpringerMaterials User Guide**

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## **3D** Molecular Structures

Within the Advanced Search page the "3D Molecules" tab opens a Speed Typing dialogue to select substances by their names, molecular formulas or CAS registry numbers. After the selection of some substance by mouse-click, Jmol, an open-source Java viewer for chemical structures in 3D, opens and presents an interactive 3D model of that molecule.

Jmol allows for a wide variety of viewer configurations and options, which are accessible via the context menu (right-hand mouse-button). Our screenshot below shows the measurement of bond lengths as an example.



Please refer to the Jmol home page, jmol.sourceforge.net, for detailed descriptions of the Jmol project, manuals and tutorials, examples, etc.

#### **Advanced Search**

Allows specified searches for Substances, Molecular Formulas, Element Systems, CAS Registry Numbers and Properties in the relevant fields. Typing effort for query formulation is reduced by suggestions of terms (Speed Typing) showing available content.

Substances field: The speed-typing list also offers CAS Registry Number and Molecular Formula next to Substance Name in brackets. Element Systems can be typed with dash (the minus key) between the individual elements. The field supports case sensitive input; please use capital letters or spaces to mark the beginning of a new element symbol.

Your Query combines all search strings from the other fields of the Advanced Search page into a Boolean query that you can either submit as is or adapt to your needs before submitting it to the search engine.

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Substances, Properties,	3D Molecules Bibliographic References Help Close	
Your Query	Go Clear	Help For Librarians Feedback
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## Bibliography

SpringerMaterials contains over 1.2 million references to primary literature (over 8000 journals referenced).

A fulltext search performed on the reference collection will immediately deliver authors, editors, publications if referenced in the database.

Typing effort for query formulation is reduced by suggestions of terms (Speed Typing) showing available content. Any prefix of occurring words can filter the suggestions efficiently as shown in the example below.

A click on one of the references leads to documents citing this literature.

_Springer Materia	<b>s</b> The Landolt-Börnstein Database	🖄 Springer
Substances, Properties, 3D	Molecules Bibliographic References Help Close	
Bibliographic References Meyer 2004 Atzeni, S., Meyer-ter-Vehn, J.: Physics Press, Oxford (2004). Keiser, D.D., Jr., Clark, C.R., Meyer, M (2004) (Crys Structure, Morphology, Pl McCutchan, E.A., Zamfir, N.V., Caprio, C., Langdown, S.D., Meyer, D.A., Rega Meyer, B., Ogriseck, S., Korobov, D.: ( Meyer, B.K., Polity, A., Farangis, B., He	Go of Inertial Confinement Fusion and High Energy Density in Matter, Clarendor K., "Phase Development in Al-Rich U-Mo-Al Alloys", Scr. Mater., 51, 893-896 nase Relations, Experimental, Interface Phenomena, 11). M.A., Casten, R.F., Amro, H., Beausang, C.W., Brenner, D.S., Hecht, A.A., H , P.H., Ressler, J.J., Yamamoto, A.D.: Phys. Rev. C 69 (2004) 024308. Gasification technologies conference, Washington, DC, October 3-6, 2004. 5, Y., Hasselkamp, D., Krämer, Th., Wang, C.: Appl. Phys. Lett. 85 (2004) 49	als
Stezowski, O.: European Physical Journ Ressler, J.J., Beausang, C.W., Ai, H., A E.A., Meyer, D.A., Regan, P.H., Sciacch	mro, H., Caprio, M.A., Casten, R.F., Hecht, A.A., Langdown, S.D., McCutchar itano, M.J.S., Yamamoto, A., Zamfir, N.V.: Phys. Rev. C 69 (2004) 034331. W., Blackmon, J.C., Chae, K., Guidry, M.W., Hix, W.R., Kozub, R.L., Lingerfe 5 A 746 (2004) 569c.	n, Go
Chemical Safety	+++ what's new +++ what's new +++ what's new +++ what's new <b>Available December 21, 2010</b> • 3 Landolt-Börnstein Volumes (mor • 3D Molecular Structure Viewer (mo • Hierarchical View for Search Result News archive	e) vre)
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## **Bookshelf Navigation**

Mirrors the organization of the Landolt-Börnstein New Book Series in Groups (I to VIII), Volumes and Subvolumes as on a bookshelf in the library. Click on one of the Groups to move to the content level, a list of available volumes will open in the main window. A click on the volume will show the Table of Contents as in the printed Landolt-Börnstein Volume. A PDF icon (<sup>1</sup>/<sub>2</sub>) shows that you have reached the content level; the adjacent "i"-icon (<sup>1</sup>/<sub>2</sub>) opens an InfoPage.

		Go Adv	anced Search			
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Nucle Group II Molec Group III Cond Group IV Physi Group V Geopl Group VI Astron Group VII Bioph Group VII Advar	nomy and physics	Group VIII: VIII/1a1 VIII/1a2 VIII/1b1 VIII/1b1 VIII/1c1 VIII/2a1 VIII/2a1 VIII/2a2 VIII/2a4 VIII/2a5 VIII/3a VIII/3a	Advanced Materials and Technologies Advanced Materials and Technologies Advanced Materials and Technologies - Advanced Materials and Technologies - Advanced Materials and Technologies - Advanced Materials and Technologies - Advanced Materials - Materials - Powde Advanced Materials - Materials - Materials - Advanced Materials - Materials - Materials - Advanced Materials and Technologies - Relations I (Thermodynamic Properties and their Relations I (Thermodynamic	Laser Physics and Application Laser Physics and Application Laser Physics and Application Laser Physics and Application Laser Physics and Application r Metallurgy Data. Metals and r Metallurgy Data. Refractory Properties of Heat Resistant S forming Data. Ferrous Alloys Energy Technologies - Nuclea Energy	s - Laser Fundamentals. s - Laser Systems. Part s - Laser Systems. Part s - Laser Applications Magnets Hard and Intermetallic steels and Superalloys Energy r Energy able Energy me A . Physical Properties an	Part 2 1 2 Materials

## **Breadcrumb Trail**

Appears horizontally at the top of InfoPages and Search Hits, is dynamic and provides a trail for the user to follow back to the starting or entry point. It is a click-able navigation and provides links back to each parent level of the current one. Here ">" serves as separator of hierarchy levels.

_Springer Materia	als The Landolt-Börnstein Database	🖄 Springer
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Particles, Nuclei and Atoms	> Advanced Technologies > Ferrous Alloys > Introduction and Definitions	
Molecules and Radicals	Introduction and Definitions	
Electronic Structure and Transport	Deformation behaviour of steel 💁 👔	
Magnetism	Flow stress of steel 🔂 👔	
Semiconductivity	Mechanical properties of steel after cold deformation 📆 👔	
Superconductivity	Young's modulus of steel 💁 🧴 Thermal conductivity of steel 💁 👔 Mechanical properties of stee	
Crystallography	Specific heat capacity of steel 1 i	el arter cold deformation
Thermodynamics	Thermal expansion of steel 💁 i Density of steel 💁 i	
Multiphase Systems		
Advanced Materials		
Advanced Technologies		
Astro- and Geophysics		
Inorganic Solid Phases 🛛 💠		
Thermophysical Properties		
Chemical Safety		
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#### Typical Breadcrumb Trail:

> Advanced Technologies > Ferrous Alloys > Introduction and Definitions

## **Chemical Safety**

Safety Documents can be accessed by clicking "Chemical Safety" on the SpringerMaterials homepage. It facilitates finding safety-relevant information on the substances included in SpringerMaterials. Substances can be specified by their proper names, molecular formulas, element systems, EINECS Numbers, or CAS-Registry Numbers.

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Molecules and Radicals				
Electronic Structure and Transport				
Magnetism	Search in			
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Superconductivity		WEEE		
Crystallography	REACH, GHS, RoHS,	WEEE		
Thermodynamics				
Multiphase Systems	o between (Webscher Terreiter (The state)		the threat and	
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Advanced Technologies			Go	
Astro- and Geophysics	REACH - Registration, Evaluation, Authorization and	Restriction	of CHemicals	
Inorganic Solid Phases 🛛 🚸	GHS - Globally Harmonized System RoHS - Restriction of Hazardous Substances			
Thermophysical Properties	WEEE - Waste from Electrical and Electronic Equipr	ment		
Chemical Safety				
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Safety data are available from REACH (Registration, Evaluation, Authorization and Restriction of Chemicals), Hazard Information (Dangerous Substances Directive 67/548/EEC), GHS (Globally Harmonized System), RoHS (Restriction of Hazardous Substances), WEEE (Waste from Electrical and Electronic Equipment) and the ECHA (European CHemicals Agency) pre-registration.

**Typical Safety Document:** 

Springer Materials	The Landolt-Börnstein Database	🖄 Springer
European regulations regarding benz	ene (C6H6)	
Name benzene	Formula: C6H6	
CAS-RN 71-43-2	Molecular Weight: 78.112 <sup>g</sup> /mol	
EG-Index: 601-020-00-8 (2004/73/EC) EINECS: 200-753-7 (EINECS2)		
lazard Information (Dangerous Subs	tances Directive 67/548/EEC)	
Hazard symbols	T F	2004/73/EC
R-Phrase	Toxic Highly flammable 45-46-11-36/38-48/23/24/25-65 R45 May cause cancer. R46 May cause heritable genetic damage. R11 Highly Flammable. R36/38 Irritating to eyes and skin. R48/23/24/25 Toxic: danger of serious damage to health by prolonged exp inhalation, in contact with skin and if swallowed.	2004/73/EC
S-Phrase	R65 Harmful: may cause lung damage if swallowed. S3-45 S53 Avoid exposure - obtain special instructions before use. S45 In case of accident or if you feel unwell, seek medical advice immedia where possible).	2004/73/EC ately (show the label
GHS classification (Globally Harmoniz Regulation on Classification, Labelling and Packa		
Signal Word Pictogram	Danger	EC/1272/2000
Hazard Statements	H225 Highly flammable liquid and vapour.	
	H225 Highly flammable liquid and vapour. H350 May cause cancer < state route of exposure if it is conclusiv other routes of exposure cause the hazard>.	ely proven that no
	H340 Maycause genetic defects <state cause="" exposure="" hazard="" if="" is="" it="" no="" of="" on="" other="" route="" routes="" the="">.</state>	onclusively proven that
	H372 Causes damage to organs <or affected,="" all="" i<br="" if="" organs="" state="">prolonged or repeated exposure <state exposure="" i<br="" of="" route="">proven that no other routes of exposure cause the hazards</state></or>	if it is conclusively
	H304 May be fatal if swallowed and enters airways.	
	H319 Causes serious eye irritation.	
	H315 Causes skin irritation.	
Precautionary Statements	P201 Obtain special instructions before use.	
	P202 Do not handle until all safety precautions have been rea	
	P210 Keep away from heat/sparks/open flames/hot surfaces.	- No smoking.
	P233 Keep container tightly closed.	
	P240 Ground/bond container and receiving equipment.	
		· · · ·
	P241 Use explosion-proof electrical/ventilating/lighting//equ	ipment.
		ipment.

## **Citation Export**

Bibliographic data of documents can be exported in the RIS format (see, e.g., "RIS (file format)" at Wikipedia), to be used with Endnote and similar programs. Click on the link "RIS-Export" in the upper right part of the InfoPages.

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#### Contents

The world's largest resource for critically evaluated physical & chemical data comprises

- the contents of the Landolt-Börnstein New Book Series (> 400 volumes) plus its electronic supplementary material,
- the LPF Multinaries Edition 2008 database on inorganic solid phases,
- a subset of the DDBST database Dortmund Data Bank Software & Separation Technology on thermophysical properties,
- Chemical Safety documents such as REACH Data.

## Context

Appears horizontally below a Search Hit and displays surroundings in which the searched string occurs. Indicated bold is whether the searched string (in blue) is found in Metadata or fulltext.

GaAs	Go Advanced Search
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4 Particles, Nuclei and Atoms 7 Molecules and Radicals	Results 1 - 10 of 987 Documents previous 12345678910 next Expanded View Clear Refine
48 Electronic Structure and Transport	Inorganic Solid Phases
15 Magnetism	Metadata - Substance: GaAs Metadata - Molecular Formula: GaAs Fulltext: GaAs, physical properties Element System: "New
473 Semiconductivity	Photoluminescence Transition in GaAs Involving D- States", Phys. Rev. Lett "Observation of a Cinnabar Phase in GaAs at High Pressure" Phys. Rev. Lett. 78 "THERMAL RECOVERY OF PHOTOCURRENT IN GaAs WITH EL2 DEFECTS AND RESIDUAL
18 Superconductivity	
90 Crystallography	Electronic Structure and Transport > Electronic Structure > Photoemission Spectroscopy > Band structure and core levels tetrahedrally-bonded semiconductors
1 Thermodynamics	GaAs 💁 🚺
21 Multiphase Systems	Metadata - Substance: GaAs Galliumarsenid (GaAs) Gallium arsenide (GaAs) Gallium monoarsenide (GaAs) Fulltext: 2.1.15 GaAs
3 Advanced Materials	(Gallium arsenide) Crystal structure: 40.47 (4) [80E1, 83K1, 84M1] Table 17. GaAs. Calculated energy eigenvalues and at 5 K [85W3 Table 18. GaAs. Theoretical and experimental 4.70 5.07 5.28 5.76 6.67 6.74 Table 19. GaAs. Energies of selected
21 Advanced Technologies	
5 Astro- and Geophysics	Semiconductivity > Semiconductors > Group IV Elements, IV-IV and III-V Compounds > Impurities and Defects > Gallium arsenide (GaAs)
156 Inorganic Solid Phases 🛛 🝕	GaAs, diffusion of Zn in GaAs 🛸 🗓
0 Thermophysical Properties	Metadata - Substance: GaAs Galliumarsenid (GaAs) gallium arsenide (GaAs) Gallium arsenide (GaAs) Gallium monoarsenide (GaAs) Metadata - Substrate: GaAs Fulltext: substance: gallium arsenide (GaAs) property diffusion of 2n in GaAs By Forum 143-147 (1997) 1117. Fig. 1. GaAs: 2n. Comparison of diffusion profiles and chemical diffusion of 2n in GaAs at T = 9000C. Curve 1: concentration
125 Chemical Safety	Semiconductivity > Semiconductors > Group IV Elements, IV-IV and III-V Compounds > Impurities and Defects > Gallium arsenide (GaAs)
	GaAs, oxygen and hydrogen in GaAs 💁 1
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	Semiconductivity > Semiconductors > Group IV Elements, IV-IV and III-V Compounds > Lattice Properties > Gallium arsenide (GaAs)
	GaAs, bulk modulus, Young's and torsional moduli 💁 🧴
	Metadata - Substance: GaAs Galliumarsenid (GaAs) Gallium arsenide (GaAs) Gallium monoarsenide (GaAs) Metadata - Substrate: GaAs Fulltext: substance: gallium arsenide (GaAs) property: bulk modulus, Young's and torsional moduli bulk modulus (GaAs-I, zincblende structure)) (in GPa) BS 98F]. For ordered GaPAs, GaInAs, and GaAsSb structures, see [90W]. pressure
	Chemical Safety European regulations regarding UPt3 / GaAs
	Metadata - Substance: UPt3 / GaAs
	Chemical Safety
	Chemical Safety European regulations regarding V3Si / GaAs

#### **Typical Context:**

Electronic Structure and Transport > Electronic Structure > Photoemission Spectroscopy > Band structure and core levels of tetrahedrally-bonded semiconductors i

GaAs 落

Metadata - Substance: GaAs ... Galliumarsenid (GaAs) ... Gallium arsenide (GaAs) ... Gallium monoarsenide (GaAs) ... Fulltext: 2.1.15 GaAs (Gallium arsenide) Crystal structure: ... 40.47 (4) [80E1, 83K1, 84M1] Table 17. GaAs. Calculated energy eigenvalues and ... at 5 K [85W3]. Table 18. GaAs. Theoretical and experimental ... 4.70 5.07 5.28 5.76 6.67 6.74 Table 19. GaAs. Energies of selected ...

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Results shown as display-optimized PDF. REACH Data and InfoPage are offered in HTML, as well as results from the LPF and DDBST database.

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Click on SpringerMaterials Logo to return to the homepage.

#### InfoPage

Provides bibliographic and metadata information for the corresponding document, and a thumbnail preview.

Bibliographic data include: How to cite the document, DOI, title, editor, author, publication date, and references. References are linked, where available, with the online versions of the original literature, usually via CrossRef or other bibliographic databases. Bibliographic data of the document can be exported in the RIS format, to be used with Endnote and similar programs.

Metadata indicate "Chapter Concepts" covered by the document, such as keywords, element systems, substances and properties.

InfoPages are marked with an "i"-icon (i) in the list of documents:

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Magnetism	Flow stress of steel 🔁 👔	
Semiconductivity	Mechanical properties of steel after cold deformation 🏂 👔	
Superconductivity	Young's modulus of steel 💁 i Thermal conductivity of steel 🛸 i	
Crystallography	Specific heat capacity of steel 📆 👔	
Thermodynamics	Thermal expansion of steel 💁 🧴 Density of steel 📆 i	
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#### Introduction and Definitions

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Plasticity of steel 🏂 👔	
Mechanical properties of steel after cold deformation 🌠 🛛	i
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Thermal expansion of steel 🏂 🛛 🚹	
Density of steel 📜 🧵	

#### Typical InfoPage:

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	, T.: <i>1 Deformation behaviour of steel</i> . Warlimont, H. (ed.). SpringerMaterials - The Landolt-Börnst <mark>germaterials.com</mark> ). Springer-Verlag Berlin Heidelberg, 2009. DOI: 10.1007/978-3-540-44760-3_2	ein Database
Title	1 Deformation behaviour of steel	Related Documents:
	<b>1 Deformation behaviour of steel</b> Introduction and Definitions M. Spittel, T. Spittel	Metal Forming Data of Ferrous Alloys -
Title In	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung,	Metal Forming Data of Ferrous Alloys - deformation behaviour
Title In Author	Introduction and Definitions M. Spittel, T. Spittel	Metal Forming Data of Ferrous Alloys -
Title In Author	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades
Title In Author Affiliation	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after
Title In Author Affiliation Part of Volume Edited by	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1</b> : Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation
Title In Author Affiliation Part of Volume Edited by	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology 2C1: Metal Forming Data of Ferrous Alloys - deformation behaviour	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel
Title In Author Affiliation Part of Volume Edited by Chapter-DOI Book-DOI	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1:</b> Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2 <b>10.1007/978-3-540-44760-3</b> (Volume in Bookshelf)	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel
Title In Author Affiliation Part of Volume Edited by Chapter-DOI	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2CI</b> : Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel 4.4 Specific heat capacity of steel
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Title In Author Affiliation Part of Volume Edited by Chapter-DOI Book-DOI	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1:</b> Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2 <b>10.1007/978-3-540-44760-3</b> (Volume in Bookshelf)	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel 4.4 Specific heat capacity of steel 4.5 Thermal expansion of steel 4.6 Density of steel
Title In Author Affiliation Part of Volume Edited by Chapter-DOI Book-DOI	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1:</b> Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2 <b>10.1007/978-3-540-44760-3</b> (Volume in Bookshelf)	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel 4.4 Specific heat capacity of steel 4.5 Thermal expansion of steel 4.6 Density of steel 5 Deformation parameters
Title In Author Affiliation Part of Volume Edited by Chapter-DOI Book-DOI	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1:</b> Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2 <b>10.1007/978-3-540-44760-3</b> (Volume in Bookshelf)	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel 4.4 Specific heat capacity of steel 4.5 Thermal expansion of steel 4.6 Density of steel 5 Deformation parameters Steel symbol/number: S235JRG2/1.0038
Title In Author Affiliation Part of Volume Edited by Chapter-DOI Book-DOI	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1:</b> Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2 <b>10.1007/978-3-540-44760-3</b> (Volume in Bookshelf)	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel 4.4 Specific heat capacity of steel 4.5 Thermal expansion of steel 4.6 Density of steel 5 Deformation parameters
In Author Affiliation Part of Volume Edited by Chapter-DOI Book-DOI Copyright	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1:</b> Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2 <b>10.1007/978-3-540-44760-3</b> (Volume in Bookshelf)	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel 4.4 Specific heat capacity of steel 4.5 Thermal expansion of steel 4.6 Density of steel 5 Deformation parameters Steel symbol/number: S235JRG2/1.0038 Steel symbol/number: S275JRC/1.0128
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Title In Author Part of Volume Edited by Chapter-DOI Book-DOI Copyright	Introduction and Definitions M. Spittel, T. Spittel Fakultät für Werkstoffwissenschaft und Werkstofftechnologie - Institut für Metallformung, Technische Universität Bergakademie Freiberg, Akademiestraße 6, 09596 Freiberg, Germany Landolt-Börnstein - Group VIII Advanced Materials and Technologies Numerical Data and Functional Relationships in Science and Technology <b>2C1</b> : Metal Forming Data of Ferrous Alloys - deformation behaviour H. Warlimont 10.1007/978-3-540-44760-3_2 10.1007/978-3-540-44760-3 (Volume in Bookshelf) Springer-Verlag Berlin Heidelberg, 2009	Metal Forming Data of Ferrous Alloys - deformation behaviour Index of steel grades 2 Flow stress of steel 3 Plasticity of steel 4.1 Mechanical properties of steel after cold deformation 4.2 Young's modulus of steel 4.3 Thermal conductivity of steel 4.4 Specific heat capacity of steel 4.5 Thermal expansion of steel 4.6 Density of steel 5 Deformation parameters Steel symbol/number: S235JRG2/1.0038 Steel symbol/number: S275JRC/1.0128 Steel symbol/number: C4D/1.0300



# Inorganic Solid Phases (LPF – Linus Pauling File Database)

The "LPF Multinaries Edition – 2008" is the most comprehensive database on inorganic solid phases. All data are normalized and standardized according to the IUCr standards. It consists of three data collections: structure and diffraction data, phase diagrams, and physical properties.

The database can be browsed and is accessible through all search functions. A logo ( $\diamondsuit$ ) next to a Search Hit indicates the source.

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Typical list of data sheets:

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0 Particles, Nuclei and Atoms 0 Molecules and Radicals	Results 1 - 9 of 9 Documents	Expanded View Clear Refine
Electronic Structure and Transport	Inorganic Solid Phases	
6 Magnetism	Al-Fe-Ti, ternary phase diagram, isothermal section 🗇	
0 Semiconductivity	Metadata - Element System: Al-Fe-Ti Fulltext: Al-Fe-Ti, ternary phase diagram, isothern Phases · phase diagrams Diagram C975736 from: Raghavan V.: "The Al-Fe-Ti (Aluminu	m-Iron-Titanium) System", Phase Diagram
0 Superconductivity	C975732 from: Raghavan V.: "The <b>Al-Fe-Ti</b> (Aluminum-Iron-Titanium) System", Phase	
1 Crystallography	Inorganic Solid Phases	
0 Thermodynamics	Al-Fe-Ti, ternary phase diagram, liquidus projection 💠	
5 Multiphase Systems	Metadata - Element System: Al-Fe-Ti Fulltext: Al-Fe-Ti, ternary phase diagram, liquidus Phases · phase diagrams Diagram C975722 from: Raghavan V.: "The Al-Fe-Ti (Aluminu	
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9 Inorganic Solid Phases		
0 Thermophysical Properties	Inorganic Solid Phases TiFeAl, crystallographic data 💠	
0 Chemical Safety	Metadata - Element System: Al-Fe-Ti Fulltext: crystallographic data Element System: A	I-Fe-Ti Inorganic Solid Phases
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#### LB (Landolt-Börnstein New Book Series)

The latest edition of this brand and the first one to be published in the English language. Started as an open series in 1961, it comprises to date > 400 volumes. To see how it is organized click Bookshelf.

[up]

#### Metadata

SpringerMaterials provides metadata for each document extracted in an editorial process: Substance, Element System, CAS Registry Number, Properties, Keywords, Main Subject, Secondary Subjects, and Bibliographic Information.

[up]

#### Navigation

SpringerMaterials offers two different views of the same content: By Subject Areas and, for aficionados of the Landolt-Börnstein New Book Series, the Bookshelf Navigation.

[up]

#### OpenSearch

Search results can be returned as an OpenSearch compliant RSS feed. The OpenSearch description document is located at http://www.springermaterials.com/content/search.xml. In order to integrate SpringerMaterials search results into your federate search, check the documentation of your search engine.

Further information about OpenSearch can be found on http://www.opensearch.org.

#### **Periodic Table Search**

Supports a search by element systems of substances and materials.

You can select elements by clicking on the symbols of the Periodic Table. Chosen elements are highlighted by an orange frame and also displayed in the central Your Selection string.

You can deselect elements by clicking on them a second time either in the Periodic Table or in the Your Selection string.

Chosen elements are highlighted by an orange frame; elements not available for further combinations are grayed-out in the Periodic Table.

Speed-typing: A list of available element systems opens. Chosen elements are marked red, black elements show further possible combinations.

Click on a possible combination from the list, available documents are shown.

To add any other search criteria, click Refine.

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-Fe-Pr-Sm-Ti	5	37 Rb	38 Sr	39 Y	40 Zr	41	42	43	44	45	46	47	48	49	50	51	52	53	54
-B-Fe-Mg-O-Ti	3	RD	sr	· ·	Zr	Nb	Mo	Тс	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Те	I	Xe
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-Ca- <mark>Fe</mark> -O-Ti-Zr	7	87 Fr	ss Ra	**	104 Rf	105 Db	105 Sq	107 Bh	108 Hs	109 Mt	110 D S	111 Rg	112 Cn	113	114		116	117	
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#### **Query Operators**

• AND

The AND operator is the default operator. If your query consists of a sequence of words separated by spaces, each space symbol is interpreted as an AND; e.g. *optical stark effect* is the same as *optical AND stark AND effect* This query will find all documents where all three words optical, stark, and effect appear somewhere as substrings in the text.

• OR

*OR* is often useful for combining closely related terms like synonyms in one query; e.g. *methylurea OR 598-50-5* will find all documents which contain at least one of the two strings *methylurea* or its CAS-Registry-Number *598-50-5*.

• Double quotes for phrase search

Double quotes around a sequence of words in a query only yields those documents containing the words in exactly this order; e.g. *"optical stark effect"* will produce only those documents containing this phrase somewhere in the text.

#### • BUT\_NOT

*BUT\_NOT* is used to specify documents by some relevant keyword but excluding some known context from the result-set; e.g. *"urea" BUT\_NOT optical* selects documents that contain the exact match of *urea* but only if it does not also contain the substring *optical*.

• { and } for nested queries

{ and } helps to handle several valid simple queries in one single and powerful nested query; e.g. {"*Ruthenium*" OR ''7440-18-8"} AND ''magnetic flux'' combines the simple query ''Ruthenium'' or alternatively ''7440-18-8'' with the required phrase ''magnetic flux''.

[up]

#### Ranking

The ranking of the displayed documents is performed according to a scoring algorithm. Relevance is calculated by location and frequency of, and conformity with the search term within the document. A hit in the Metadata is scored higher than one occurring in the fulltext. Exact matches are preferred over substring matches.

# REACH

The Chemical Safety Search finds REACH-relevant (Registration, Evaluation, Authorization and Restriction of CHemicals) information on the substances (alternatively CAS-Registry Numbers, Molecular Formula) included in SpringerMaterials. Also described, where applicable, are the GHS (Globally Harmonized System), RoHS (Restriction of Hazardous Substances), and WEEE (Waste from Electrical and Electronic Equipment).

[up]

## Refine

To select or deselect subject areas, or to add any other search criteria, click Refine. You will be automatically directed to the Advanced Search, where you can narrow down your results with more specialized queries; then click Go.

# **Sample Searches**

#### **Simple Search**

The Simple Search field is found in the center of the SpringerMaterials homepage and replicated in the field below the SpringerMaterials logo. Typing in a query is the quickest way to find data; however, to get a more precise result, refinement is possible in a second step.

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Home Bookshelf	Periodic Table		Help	For Librarians	Feedback
Particles, Nuclei and Atoms					
Molecules and Radicals					
Electronic Structure and Transport					
Magnetism		Search in	1		
Semiconductivity					
Superconductivity					
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Thermodynamics	-	<u>_</u> p			
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Advanced Technologies		rmicacid	60		
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Inorganic Solid Phases 🛛 🚸	250,000 Substance	ormic acid d		ons	
Thermophysical Properties		ormic acid-d			
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		ormic hydrazide			
	F	ormic acid dimer			
	<	previous Page 1 of	21	next >	
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In the example we are interested in all data available about *"formic acid"*. Typing in the first characters opens a list of suggestions (via "Speed Typing") which shows the available content. A click on the first term and pushing the "Go" button executes the query. A list of available documents is shown.

In the default view of search results, each document found by the query is presented by indicating the path to the document in the systematic hierarchy via the breadcrumb trail, the title of the document, and the context of the search terms within the document:

Springer Materia	als The Landolt-Börnstein Database 🖉 Spri	nger
"formic acid"	Go Advanced Search	
Home Bookshelf	Periodic Table For Librarians	Feedbac
0 Particles, Nuclei and Atoms	Results 1 - 10 of 567 Documents previous 12345678910 next Expanded View Cle	ar Refine
127 Molecules and Radicals		Kentile
26 Electronic Structure and Transport	Molecules and Radicals > Molecular Structure > Organic Molecules > One or Two Carbon Atoms > Element sy	stems C-H2
17 Magnetism	Metadata - Substance: formic acid Formic acid CH2O2 (formic acid) Fulltext: 228 CH2O2 Formic acid Cs MW r a) C-H	I C=0 C-0 O-
0 Semiconductivity	Å 1.0981(16)	
0 Superconductivity	Thermophysical Properties	
18 Crystallography	Formic acid P	
324 Thermodynamics	Metadata - Substance: Formic acid Fulltext: Formic acid Thermophysical Data in the Dortmund Number Name 1 CH2O2 4	6.026 64-18-
0 Multiphase Systems	Formic acid List of Available Properties Pure	
3 Advanced Materials	Thermodynamics > Mechanical Properties > Surface Tension > Pure Liquids and Binary Liquid Mixtures	
1 Advanced Technologies	Data on pure liquids 💁 🧴	
0 Astro- and Geophysics	Metadata - Substance: formic acid CH2O2 (formic acid) formic acid ethyl ester formic acid hexyl ester Fulltext: 30.4 24.87 2002 C11H18O2 formic acid trans-3,7-dimethyl-oct-2,6-dien-1-yl	86 30.81 26.9
0 Inorganic Solid Phases 🛛 💠	Thermodynamics > Thermodynamical Properties > Organic Compounds > Vapor Pressure and Antoine Const.	ants > Oxyge
31 Thermophysical Properties 🗦 🍃	Containing Organic Compounds	unio – okyge
	compounds C2C8 💁 🧴	
20 Chemical Safety	Metadata - Substance: formic acid CH2O2 (formic acid) benzoyl formic acid	
	Thermodynamics > Mechanical Properties > Viscosity > Organometallic Liquids and Binary Liquid Mixtures > N organic compounds Part 1 💈 i	1ixtures of
	Metadata - Substance: formic acid CH2O2 (formic acid) formic acid ethyl ester Formic acid, ethyl ester C3H6O2 (fo ester) Fulltext: 0.1669 0.1944 0.2394 498 CH2O2 (1) formic acid *) 64-18-6 CH3NO (2) formamide 75-12-7 3.196 3.341 499 CH2O2 (1) formic acid *) 64-18-6 CH4O (2) methanol 67-56-1 T/	
	Electronic Structure and Transport > Dielectricity > Pure Liquids and Binary Liquid Mixtures: Dielectric Consta Pure Liquids: References 1 i	nts
	Metadata - Substance: formic acid CH2O2 (formic acid) formic acid ethyl ester	
	Thermodynamics > Electrical Properties > Electrochemistry: Thermodynamics and Kinetics > Electrode potent Table: Metal electrode potentials in nonaqueous electrolyte systems 1	tials

A more compact list not showing the context can be obtained by clicking "Compact View":

SpringerMateria	als The Landolt-Börnstein Database	🖄 Springer									
"formic acid"	Go Advanced Search										
Home Bookshelf	Periodic Table He	elp For Librarians Feedbac									
0 Particles, Nuclei and Atoms	Results 1 - 10 of 567 Documents previous 12345678910 next	Compact View Clear Refine									
127 Molecules and Radicals	Results 1 - 10 of 507 Documents previous 12 54 507 0 910 mext										
26 Electronic Structure and Transport	Molecules and Radicals > Molecular Structure > Organic Molecules > One or Two	Expanded View Hierarchical View									
17 Magnetism											
0 Semiconductivity	Thermophysical Properties										
0 Superconductivity	Formic acid P										
18 Crystallography	Thermodynamics > Mechanical Properties > Surface Tension > Pure Liquids and Binary Liquid Mixtures										
324 Thermodynamics	Data on pure liquids Table 1										
0 Multiphase Systems											
3 Advanced Materials	Thermodynamics > Thermodynamical Properties > Organic Compounds > Vapor Pressure and Antoine Constants > Ox Containing Organic Compounds										
1 Advanced Technologies	compounds C2C8 1										
0 Astro- and Geophysics											
0 Inorganic Solid Phases 🛛 🚸	Thermodynamics > Mechanical Properties > Viscosity > Organometallic Liquids a organic compounds	nd Binary Liquid Mixtures > Mixtures of									
31 Thermophysical Properties	Part 1 🔁 🚺										
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20 Chemical Safety	Electronic Structure and Transport > Dielectricity > Pure Liquids and Binary Liquid Pure Liquids: References 1	d Mixtures: Dielectric Constants									
	Thermodynamics > Electrical Properties > Electrochemistry: Thermodynamics an Table: Metal electrode potentials in nonaqueous electrolyte systems										
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	Molecules and Radicals > NMR Spectroscopy > Oxygen-17: Shifts and Coupling C bounds (C=O)	Constants > Compounds with C-O double									

The "Hierarchical View" of the query results locates search hits within the Subject Areas and the Navigation Tree. This allows an easy location of search hits within a specific context:

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formic acid	Go Advanced Search	
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0 Particles, Nuclei and Atoms	> Search Results	
198 Molecules and Radicals	Search Results	Hierarchical View Clear Refine
95 Electronic Structure and Transport	(198) Molecules and Radicals (11) Molecules, General Topics	Expanded View
35 Magnetism	(75) Molecular Constants (40) Molecular Structure	Compact View
0 Semiconductivity	(19) NMR Spectroscopy	
0 Superconductivity	(32) NQR Spectroscopy (13) Radicals, Magnetism	
28 Crystallography	(8) Radicals, Reaction Kinetics (95) Electronic Structure and Transport	
476 Thermodynamics	(44) Dielectricity	
0 Multiphase Systems	(51) Optics (35) Magnetism	
5 Advanced Materials	<ul> <li>(24) Diamagnetic Susceptibilities</li> <li>(11) Coordination and Organometallic Compounds</li> </ul>	
1 Advanced Technologies	(28) Crystallography (22) Crystal Structure	
0 Astro- and Geophysics	(1) Atomic Defects	
1 Inorganic Solid Phases 🛛 💠	<ul> <li>(5) Solid Surfaces</li> <li>(476) Thermodynamics</li> <li>(282) Thermodynamical Properties</li> </ul>	
31 Thermophysical Properties 📿	(190) Mechanical Properties (4) Electrical Properties	
26 Chemical Safety	(5) Advanced Materials (1) Nucleic Acids (4) Proteins	
	(1) Advanced Technologies (1) Energy	
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Clicking "Refine" opens the "Advanced Search" form.

## **Advanced Search**

The Advanced Search allows specific searches for chemical substances and their properties. Chemical substances can be specified by their proper names, molecular formulas, element systems, or CAS registry numbers.

Springer Materials The Landolt-Börnstein Database	er
Substances, Properties, 3D Molecules Bibliographic References Help Close	
Your Query Help For Librarians F	eedback
"formic acid" {"surface tension"}	Refine
Search for Substances / Molecular Formulas / Element Systems / CAS Registry Numbers 228 CH2O2 Formic acid Cs MW r a) C-H C=O	
Properties	
"surface tension"	
surface twist         and the surface tension           Search in surface target thickness         Surface tension	564-18-6
Particles surface phase transition s and Binary Liquid Mixtures	
Moleculi nuclear surface thickness	
Magneti Surface termination dependence	.81 26.99
Semicol     Previous     Page 1 of 1     next >       Supercoll     Crystallography     exactly this phrase       Thermodynamics       Multiphase Systems     but none of these words	≻ Oxygen
Advanced Technologies     Advanced Technologies     Astro- and Geophysics	es of
Inorganic Solid Phases     Formic acid, ethyl ester C3H6O2 (formic a     H3NO (2) formamide 75-12-7 3.196 3.286     Chemical Safety	<b>cid</b> ethyl 5 3.314
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Pure Liquids: References 📓 🧴 Metadata - Substance: formic acid CH2O2 (formic acid) formic acid ethyl ester	
Thermodynamics > Electrical Properties > Electrochemistry: Thermodynamics and Kinetics > Electrode potentials Table: Metal electrode potentials in nonaqueous electrolyte systems a i	
Metadata - Substrate: formic acid Fulltext: ethanol, Cd(NO3)2, NH4Br Cd=Cd2++2e=, formic acid, CdCl2 Cd=Cd2+, formamide KCl methanol, CaCl2 Ca=Ca2++2e-, formic acid, Ca formate Ca=Ca2++2e-, propylene Cs=Cs++e-, ammonia, CsNO3 Cs=C formic acid, 0.01 M CsHCOO Cs=Cs++e-, acetonitrile, ethanol, NH4NO3 Pb=Pb2++2e-, formic acid, Pb(HCOO)2 Pb=Pb2++2 anhydrous	Cs++e-,
Molecules and Radicals > Molecules, General Topics > Molecular Acoustics > Sound absorption and dispersion in ga liquids > Chemically homogeneous liquids	ises and
The frequency-independent portions of sound absorption in organic liquids 🔹 🗓 Metadata - Substance: formic acid CH2O2 (formic acid) formic acid butyl ester	

In the example the user is searching for the surface tension of formic acid. Thus, "*formic acid*" was typed in the substance search field and the appropriate entry was chosen from the list of suggestions. Then the first characters of the property "*surface tension*" were typed and the corresponding hit was chosen from the list of suggestions. "Your Query" combines all search strings from the other fields of the advanced search page into a Boolean query that can be submitted as is or adapted if necessary. Pushing the "Go" button executes the query.

The following screenshot shows the list of available documents:

Springer Materia	als The Landolt-Börnstein Database	🖉 Springer
rmic acid" {"surface tension"}	Go Advanced Search	
Home Bookshelf	Periodic Table Help	For Librarians Feedba
0 Particles, Nuclei and Atoms	Results 1 - 6 of 6 Documents	Expanded View Clear Refine
0 Molecules and Radicals		Cieal Keine
0 Electronic Structure and Transport	Thermophysical Properties	
0 Magnetism	Metadata - Substance: Formic acid Metadata - Property: Surface Tension Fulltext: Formic a	rid Thermonbusical Data in the Dortmund
0 Semiconductivity	Pure Component Properties (Surface Tension) Pure Component Properties (Thermal	
0 Superconductivity	Thermodynamics > Mechanical Properties > Surface Tension > Pure Liquids and Bir	ary Liquid Mixtures
0 Crystallography	Data on pure liquids 1 i	hary Eigene Mixteres
5 Thermodynamics	Metadata - Substance: formic acid CH2O2 (formic acid) Metadata - Property: surface tens	ion
0 Multiphase Systems		
0 Advanced Materials	Thermodynamics > Mechanical Properties > Surface Tension > Pure Liquids and Bir IV/16)	hary Liquid Mixtures (Supplement to
0 Advanced Technologies	Surface tension of the mixture (1) water; (2) formic acid 🛸 🤳	
0 Astro- and Geophysics	Metadata - Substance: formic acid CH2O2 (formic acid) Metadata - Property: surface tens binary liquid mixtures Fulltext: 2 Hill Formula H2O CH2O2 Name water formic acid CAS Nun Table 1. H2O (1): CH2O2 (2). Surface Tension at T= 293.15 K (1 997ALVI). x 2 0.000	
0 Inorganic Solid Phases 🛛 💠		
1 Thermophysical Properties 🗦 ⊃	Thermodynamics > Mechanical Properties > Surface Tension > Pure Liquids and Bir IV/16)	nary Liquid Mixtures (Supplement to
0 Chemical Safety	Surface tension of formic acid 🔼 🧴	
	Metadata - Substance: formic acid CH2O2 (formic acid) Metadata - Property: surface tens liquids Fulltext: Substance Hill Formula CH2O2 Name formic acid CAS Number 64-18-6 Ok	
	Thermodynamics > Mechanical Properties > Surface Tension > Pure Liquids and Bir	nary Liquid Mixtures
	Data on mixtures 💁 🧴	
	Metadata - Substance: formic acid Metadata - Property: surface tension	
	Thermodynamics > Mechanical Properties > Surface Tension > Pure Liquids and Bir IV/16) Index of Substances i	nary Liquid Mixtures (Supplement to

The document of the first hit contains data on the surface tension of formic acid from the database on Thermophysical Properties. The second hit provides a PDF document from a Landolt-Börnstein compilation on surface tension covering formic acid. Further hits locate data about mixtures of formic acid with other liquids, e.g. with water.

A combined substance/property search is a typical use case for SpringerMaterials.

The Advanced Search page is opened by clicking on the "Advanced Search" button or by clicking the "Refine" button in the list of hits, e.g., as a second step after a simple search.

Besides the search for chemical substances and properties, the "Advanced Search" allows you to search for a specific word, for exact phrases, and to exclude documents containing specific words from the search results. Moreover, search can be restricted to one or more subject areas.

## **Bibliography Search**

The "Bibliography Search" is part of the Advanced Search feature. SpringerMaterials contains over 1 million references to primary literature (over 8000 journals are referenced).

Springer Materials The Landolt-Börnstein Database	🖄 Springer
Substances, Properties, 3D Molecules Bibliographic References Help Close	Por Librarians Feedback
Search for	
Bibliographic References	
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A.K. Pradhan and J. Peng, in Analysis of Emission Lines, Ed. R.E. Williams and M. Livio, Cambridge University Press (1995).	
Aaronson, M. in: Infrared Astronomy, Int. Astron. Union Symp. 96 (Wynn-Williams, C.E., Cruitshank, D.P., eds.), Reidel, Dordrecht (1981) p. 297.	
Ababio, B.D., McElroy, P.J., Salt, B., Williamson, A.G.: Chem. Eng. J. (Lausanne) 47 (1991) 113.	
Ababio, B.D., McElroy, P.J., Williamson, A.G.: Fluid Phase Equilibria 95 (1994) 329.	als
Abegg, R., Hutton, J.D., Williams-Norton, M.E.: Nuclear Physics A 303 (1978) 121.	415
Abel, E.W., Bhargava, S.K., Bhatti, M.M., Kite, K., Mazid, M.A., Orrell, K.G., Sik, V., Williams, B.L., Hursthouse, M.B., Abdul Malik, K.M.: J. Chem. Soc., Dalton Trans. (1982) 2065.	
Abrahams, S.C., Bernstein, J.L., Sherwood, R.C., Wernick, J.H., Williams, H.J.: J. Phys. Chem. Solids 25 (1964) 1069.	
Abrahams, S.C., Williams, H.J.: J. Chem. Phys. 39 (1963) 2923-2933.	Go
Adam, W., Sahin, C., Sendelbach, J., Walter, H., Chen, GF., Williams, F.: J. Am. Chem. Soc. 116 (1994) 2576.	a in Materials Science
Adam, W., Walter, H., Chen, GF., Williams, F.: J. Am. Chem. Soc. 114 (1992) 3007.	0,000 Literature Citations
Addinall, R., Murray, R., Newman, R.C., Wagner, J., Parker, S.D., Williams, R.L., Droopad, R., De Oliviera, A.G., Ferguson, I., Stradling, R.A.: Semicond. Sci. Technol.6 (1991) 147.	
Addinall, R., Murray, R., Newman, R.C., Wagner, J., Parker, S.D., Williams, R.L., Droopad, R., De Oliviera, A.G., Ferguson, I., Stradling, R.A.: Semiconductor Science and Technology 6 (1991) 147.	
Substances, Properties,       3D Molecules       Bibliographic References       Help       Close         Search for       Example       Go       Help       For Librari         Bibliographic References       Williams       Go       AK. Pradhan and J. Peng, in Analysis of Emission Lines, Ed. R.E. Williams and M. Livio, Cambridge University Press (1995).       Aronson, M. in: Infrared Astronomy, Int. Astron. Union Symp. 96 (Wynn-Williams, C.E., Cruitshank, D.P., eds.), Reidel, Dordrecht (1991) p. 297.       Ababio, B.D., McElroy, P.J., Salt, B., Williamson, A.G.: Chem. Eng. J. (Lausanne) 47 (1991) 113.       Ababio, B.D., McElroy, P.J., Sult, B., Williamson, A.G.: Chem. Eng. J. (Lausanne) 47 (1991) 113.       Ababio, B.D., McElroy, P.J., Williams-Norton, M.E.: Nuclear Physics A 303 (1978) 121.       Ababio, B.D., McElroy, P.J., Williams-Norton, M.E.: Nuclear Physics A 303 (1978) 121.         Ababio, B.D., McElroy, P.J., Williams, S.C., Wernick, J.H., Williams, H.J.: J. Phys. Chem. Solids 25 (1964) 1069.       Abadu Malik, K.M.: J. Chem. Drass. (1982) 2025.         Adarn, W., Sahin, C., Sendelbach, J., Walter, H., Chen, GF., Williams, F.: J. Am. Chem. Soc. 116 (1994) 2576.       Adam, W., Walter, H., Chen, GF., Williams, F.: J. Am. Chem. Soc. 116 (1994) 2576.         Adam, W., Walter, H., Chen, GF., Walgner, J., Parker, S.D., Williams, R.L., Droopad, R., De Oliviera, A.G., Ferguson, T., Stradling, R.A.: Semiconductor Science and Technology 6 (1991) 147.       Mins, H., Williams, J.L.: J. Org. Chem. 17 (1952) 900.         Adlinall, R., Murray, R., Newman, R.C., Wagner, J., Parker, S.D., Williams: Journal of the Chemical Society 1963, 4649.	
Allen, E.A., B.J. Brisdon, D.A. Edwards, G.W.A. Fowles and R.G. Williams: Journal of the Chemical Society 1963, 4649.	
< previous Page 1 of 119 next :	>
News archive	
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A fulltext search performed on the reference collection will immediately deliver authors, editors, and publications referenced in the database. Typing effort for query formulation is reduced by suggestions of terms ("Speed Typing") showing available content. In this example we typed in *"williams"* and get suggestions of possible references where the substring "williams" occurs. A click on one of the references leads to documents citing this literature.

#### **Chemical Safety Document Search**

_Springer Materia	<b>ls</b> The Land	olt-Börnstein Database	🖄 Springer
	Go Advance	d Search	S, ROHS, WEEE Element Systems / CAS Registry Numbers Go Authorization and Restriction of CHemicals term Substances
Home Bookshelf	Periodic Table		Help For Librarians Feedback
Particles, Nuclei and Atoms			
Molecules and Radicals			
Electronic Structure and Transport			
Magnetism		Search in	
Semiconductivity			
Superconductivity			
Crystallography		REACH, GHS, ROHS	, WEEE
Thermodynamics			
Multiphase Systems		Outstands (Malesular Especials (Element Outsta	- / CAR Basister Murchan
Advanced Materials		Substances / Molecular Formulas / Element Syster	
Advanced Technologies			
Astro- and Geophysics		REACH - Registration, Evaluation, Authorization a	nd Restriction of CHemicals
Inorganic Solid Phases 🛛 💠		GHS - Globally Harmonized System RoHS - Restriction of Hazardous Substances	
Thermophysical Properties		WEEE - Waste from Electrical and Electronic Equ	ipment
Chemical Safety			
3 Springer 2010	Iroprint   D	artners   Contact   Disclaimer   System Requiremen	ts – – – – – – – – – – – – – – – – – – –

The "Chemical Safety Document Search" can be accessed by clicking "Chemical Safety" on the SpringerMaterials home page. It facilitates finding safety-relevant information on the substances included in SpringerMaterials. Substances can be specified by their proper names, molecular formulas, element systems, or CAS-Registry Numbers.

Springer Materials The Landolt-Börnstein Database 🖉 Sp		🖄 Sprin	nger			
		Go Advanc	ed Search			
Home Bookshe	əlf	Periodic Table		Help	For Librarians	Feedback
Particles, Nuclei and Atoms						
Molecules and Radicals						
Electronic Structure and Transpo	ort					
Magnetism			Search in			
Semiconductivity						
Superconductivity					-	
Crystallography			REACH, GHS, RoH	S, WEEE		
Thermodynamics						
Multiphase Systems						
Advanced Materials			Substances / Molecular Formulas / Element Syst	ems / CAS Reg		
Advanced Technologies			benzene		Go	
Astro- and Geophysics			benzene (C6H6) benzene-d6 (C6D6)		emicals	
Inorganic Solid Phases	4		Benzene-1,4-d2 (C6H4D2)			
Thermophysical Properties	ž		benzenemethanol (C7H8O)			
Thermophysical Propercies	-		Cyanato-benzene (C7H5NO)			
Chemical Safety			benzene selenol (C6H6Se)			
			butoxy-benzene (C10H14O)			
			dimethyl-benzene (C <sub>8</sub> H <sub>10</sub> )			
			propoxy-benzene (C9H12O)			
			Benzene, hexaiodo- (C6I6)			
			< previous Page 1 of 67		next >	
© Springer 2010		Imprint	Partners   Contact   Disclaimer   System Requirem		Doworod	by Informatik I

"Chemical Safety Document Search" finds data from REACH (Registration, Evaluation, Authorization and Restriction of Chemicals). If available, data about Hazard Information (Dangerous Substances Directive 67/548/EEC), GHS (Globally Harmonized System), RoHS (Restriction of Hazardous Substances), WEEE (Waste from Electrical and Electronic Equipment) and on the European CHemicals Agency (ECHA) pre-registration are also given.

In the example the user is interested in REACH-relevant data of *benzene*. A click on the first entry in the list of suggestions opens the corresponding data sheet:

_Springer Materials	The Lando	olt-Börnstein Database 🖄 Sp	ringer
European regulations regarding benze			
Name benzene	Form	ula: C6H6	
CAS-RN 71-43-2	Molec	ular Weight: 78.112 <sup>g</sup> / <sub>mol</sub>	
EG-Index: 601-020-00-8 (2004/73/EC) EINECS: 200-753-7 (EINECS2)			
Hazard Information (Dangerous Substa	ances Directive 67	/548/EEC)	
Hazard symbols			2004/73/EC
	T Toxic	F Highly flammable	2004/73/EC the label EC/1272/2008 that no proven that ough isively
R-Phrase	45-46-11-36/3 R45 May cause R46 May cause R11 Highly Fla	8-48/23/24/25-65 cancer. s heritable genetic damage.	2004/73/EC
	R48/23/24/25	Toxic: danger of serious damage to health by prolonged exposure through	
		ontact with skin and if swallowed. may cause lung damage if swallowed.	
S-Phrase	53-45	nay baase lang aamage it swallowea.	2004/73/EC
		osure - obtain special instructions before use. : accident or if you feel unwell, seek medical advice immediately (show the label -).	
GHS classification (Globally Harmonize Regulation on Classification, Labelling and Packag	e <b>d System)</b> ing of Substances and M	Mixtures (CLP)	
Signal Word	Danger		EC/1272/2008
Pictogram	*		
Hazard Statements	H225	Highly flammable liquid and vapour.	
	H350	May cause cancer <state cause="" conclusively="" exposure="" hazard="" if="" is="" it="" no="" of="" other="" proven="" route="" routes="" that="" the="">.</state>	
	H340	Maycause genetic defects <state conclusively="" exposure="" if="" is="" it="" of="" proven="" route="" that<br="">no other routes of exposure cause the hazard&gt;.</state>	:
	H372	Causes damage to organs <or affected,="" all="" if="" known="" organs="" state=""> through prolonged or repeated exposure <state conclusively<br="" exposure="" if="" is="" it="" of="" route="">proven that no other routes of exposure cause the hazard&gt;.</state></or>	
	H304	May be fatal if swallowed and enters airways.	
	H319	Causes serious eye irritation.	
	H315	Causes skin irritation.	
Precautionary Statements	P201	Obtain special instructions before use.	
	P202 P210	Do not handle until all safety precautions have been read and understood.	
	P210 P233	Keep away from heat/sparks/open flames/hot surfaces No smoking. Keep container tightly closed.	
	P240	Ground/bond container and receiving equipment.	
	P241	Use explosion-proof electrical/ventilating/lighting//equipment.	
	P242	Use only non-sparking tools.	
11	P243	Take precautionary measures against static discharge.	l

## **Periodic Table Search**

A click on the button "Periodic Table" opens a window showing the Periodic Table of Chemical Elements.

		Go		dvance	d Sear	ch														
Home Bookshelf		Period	ic Tabl	le 💧									He	lp	Fo	r Libra	rians	F	eedb	a
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	1	Ĥ	D D	Ť		Colori			alialia	a		hala							He	
	2	Li	Be     Select elements by clicking on the symbols.     Deselect elements by clicking a second time.     Your Selection									B	6 C	7 N	ů	° F	<sup>10</sup> Ne			
	з	na <sup>11</sup>	<sup>12</sup> Mg			Your	Select	tion						AI	si Si	15 P	16 S	с <sup>17</sup> СІ	18 Ar	
	4	19 K	20 Ca	21 Sc	22 Ti	23 ¥	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 A s	34 Se	35 Br	36 Kr	
	5	37 Rb	38 Sr	39 ¥	2r Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	es Pd	47 Ag	49 C d	₄s In	so Sn	sı Sb	52 Te	53 I	54 Xe	
	6	55 C s	56 Ba	*	72 Hf	73 Ta	74 ₩	75 Re	76 Os	77 Ir	78 Pt	79 Au	eo Hg	81 TI	82 Pb	83 Bi	84 Po	as At	as Rn	
	7	87 Fr	88 Ra	**	<sup>104</sup> Rf	105 Db	105 Sg	107 Bh	108 Hs	109 Mt	110 Ds	nii Rg	112 Cn	113	114	115	116	117	118	
			*	57 La	58 Ce	s≋ Pr	60 Nd	61 Pm	e₂ Sm	a Eu	64 Gd	≝ Tb	66 D y	67 Ho	69 Er	ee Tm	70 Yb	71 Lu		
			<b>4</b> .4	SS AC	90 Th	91	92 U	93 Np	94	95	96 Cm	97 Bk	99	99	100	101 Md	102 NO	103 Lr		
			**	AC	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	NO	Lr		

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It supports a search by element systems of substances and materials. Elements can be selected by clicking on the symbols of the Periodic Table. The elements chosen can be deselected by clicking on them a second time either in the Periodic Table or in the "Your Selection" string.

		G		dvance	d Sear	rch														
		_			u sear	CIT														
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I-Cr-Fe-Ge		1 IA	2 11A	3 111B	4 IVB	vB	VIB	VIIB	VIIIB	9 VIIIB	VIIIB	11 IB	12 IIB	13 111A	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA	
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-Cr-Fe-O-Sr		87	83		104	105	105	107	108	109	110	111	112	113	114	115	116	117	118	i.
I-Cr-Fe-O-Ti	7	Fr	Ra	**	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn							L
I-Cr-Fe-O-Zn																				1
I-B-Co <mark>-Cr-Fe</mark> -Nd			*	57 La	sa Ce	se Pr	60 Nd	61 Pm	Sm	ea Eu	Gd Gd	Tb	66 D y	67 Ho	68 Er	Tm	70 Yb	71 Lu		
I-Ca-Cr-Fe-O-Si				La			i nu		5111	Lu	du	10	07	110			10	Lu		
I-Co-Cr-Fe-O-Zn			**	ee Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	Am	95 Cm	97 Bk	98 Cf	99 Es	100 Em	101 Md	102 No	103 Lr		
I-Cr-Fe-Ga-Ni-O						1.4		i de	1.4			DR		2.5		110	110		i i	
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Al-Cr-Fe-Nb-Ni-Ti																				

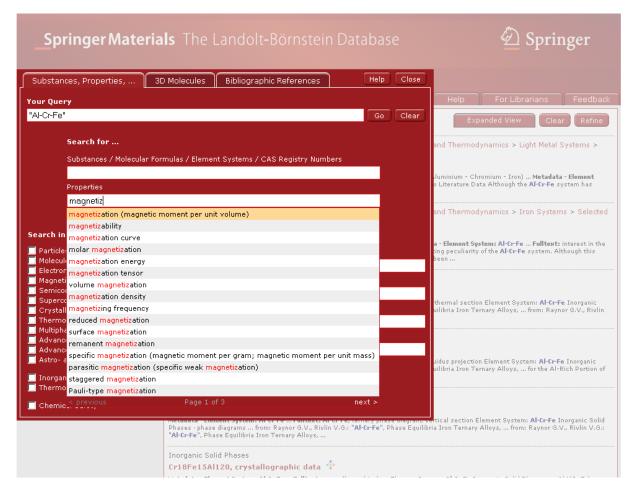
Elements chosen are highlighted by an orange frame and are also displayed in the central "Your Selection" string. Elements not available for further combinations are grayed-out in the Periodic Table.

After choosing an element a list of available element systems opens on the left hand. Chosen elements are marked red, black elements show further possible combinations. Click on a possible combination from the list and a list of available documents is shown in a new window.

In this example, we have chosen aluminum (*Al*), chromium (*Cr*) and iron (*Fe*). By clicking the first entry in the list of available element systems (''*Al-Cr-Fe*''), a list of documents containing this element system appears. Search hits are from Landolt-Börnstein documents and the database on Inorganic Solid Phases, as indicted by the icons.

_Springer Materi	als The Landolt-Börnstein Database	🖄 Springer						
Al-Cr-Fe"	Go Advanced Search							
Home Bookshelf	Periodic Table He	lp For Librarians Feedba						
0 Particles, Nuclei and Atoms	Results 1 - 10 of 34 Documents previous 1234 next	Expanded View Clear Refine						
0 Molecules and Radicals	Results 1 - 10 01 34 Documents previous 12 34 next	Clear Kenne						
1 Electronic Structure and Transport	Multiphase Systems > Ternary Alloys > Phase Diagrams, Crystallography and Th Aluminum (Al-X-Y) Ternary Alloys	ermodynamics > Light Metal Systems >						
15 Magnetism	Al-Cr-Fe 💁 🧵							
0 Semiconductivity	Metadata - Substance: Al-Cr-Fe Al-Cr-Fe (Aluminium-Chromium-Iron) Al-Cr-Fe (Aluminiu System: Al-Cr-Fe Fulltext: Al-Cr-Fe Aluminium - Chromium - Iron Gautam Sidorko Litera							
0 Superconductivity	undergone many investigations, the Al-Cr-Fe equilibrium diagram has not been	ture Data Anthough the Arthree system has						
3 Crystallography	Multiphase Systems > Ternary Alloys > Phase Diagrams, Crystallography and Th	armadupamias a Iran Systems a Salaste						
0 Thermodynamics	Systems from Al-B-Fe to C-Co-Fe	erniodynamics > from systems > selecte						
6 Multiphase Systems	Aluminium-Chromium-Iron 💁 🗓							
0 Advanced Materials	Metadata - Substance: Al-Cr-Fe Al-Cr-Fe (Aluminium - Chromium - Iron) Metadata - Element System: Al-Cr-Fe Fulltext: interest in t phase relations in the Al-Cr-Fe system, particularly as alloys of this also an interesting peculiarity of the Al-Cr-Fe system. Although this							
0 Advanced Technologies	system has undergone many investigations, the Al-Cr-Fe equilibrium diagram has not been	anancy of the Al of Te system. Although this						
0 Astro- and Geophysics	Inorganic Solid Phases							
9 Inorganic Solid Phases 🛛 👌								
0 Thermophysical Properties	Metadata - Element System: Al-Cr-Fe Fulltext: Al-Cr-Fe, ternary phase diagram, isothermal Solid Phases · phase diagrams from: Raynor G.V., Rivlin V.G.: "Al-Cr-Fe", Phase Equilibria V.G.: "Al-Cr-Fe", Phase Equilibria Iron Ternary Alloys,							
0 Chemical Safety	Inorganic Solid Phases							
	Al-Cr-Fe, ternary phase diagram, liquidus projection 💠							
	Metadata - Element System: Al-Cr-Fe Fulltext: Al-Cr-Fe, ternary phase diagram, liquidus pr Solid Phases · phase diagrams from: Raynor G.V., Rivlin V.G.: <b>"Al-Cr-Fe</b> ", Phase Equilibria : the Al-Cr-Fe Phase Diagram", Z. Metallkd. 78 (1987)							
	Inorganic Solid Phases							
	Al-Cr-Fe, ternary phase diagram, vertical section 💠							
	Metadata - Element System: Al-Cr-Fe Fulltext: Al-Cr-Fe, ternary phase diagram, vertical se Phases - phase diagrams from: Raynor G.V., Rivlin V.G.: "Al-Cr-Fe", Phase Equilibria Iron T "Al-Cr-Fe", Phase Equilibria Iron Ternary Alloys,							
	Inorganic Solid Phases Cr18Fe15Al120, crystallographic data 💠							

To add any other search criteria, a click on the "Refine" button (top right) opens the Advanced Search window:



In this case we are interested in the magnetization of the element system Al-Cr-Fe. Typing in the first characters into the Properties search field opens the speed-typing list, and the entry "magnetization" can be chosen. The complete search phrase will then be shown in the field "Your Query" which can be edited or left as it is. Pushing the "Go" button will lead to the list of available documents.

## Search

SpringerMaterials offers Simple Search, the Periodic Table Search and the Advanced Search.

A query is case insensitive and substring matching in its basic form. E.g., *crystal* as a query produces the same results as *Crystal* or *CRYSTAL* and also returns hits in *crystalline*, *nanocrystalline*, etc.

Typing effort for query formulation is reduced by suggestions of terms (Speed Typing) showing available content.

You can either type your query, then click "Go" or select a term from the speed-typing list and click "Go". While the suggestions are specific for the input fields, e.g., Molecular Formulas are supplied, the selection of any of the suggestions is not required; in fact any string can be placed in any of the fields to perform a search.

#### **Search Hit**

Each Search Hit shows the following three lines from top to bottom: Breadcrumb Trail, Fulltext Document, Context (LB, LPF, DDBST, or Chemical Safety).

Thermophysical Properties Methanol 🍛 Metadata - Substance: Methanol ... Metadata - Property: Yapor Pressure ... Fulltext: Methanol Thermophysical Data in the Dortmund ... Pure Component Properties (Vapor Pressure) Pure Component Properties (Dynamic ...

Thermodynamics > Thermodynamical Properties > Organic Compounds > Vapor Pressure and Antoine Constants > Oxygen Containing Organic Compounds i.

compounds C2...C8 🚺

Metadata - Substance: Methanol ... furanmethanol ... Oxiranmethanol ... Metadata - Property: vapor pressure ... Fulltext: Inorganic Compounds 2 Tabulated Data on Vapor Pressure of Oxygen Containing Organic Compounds ...

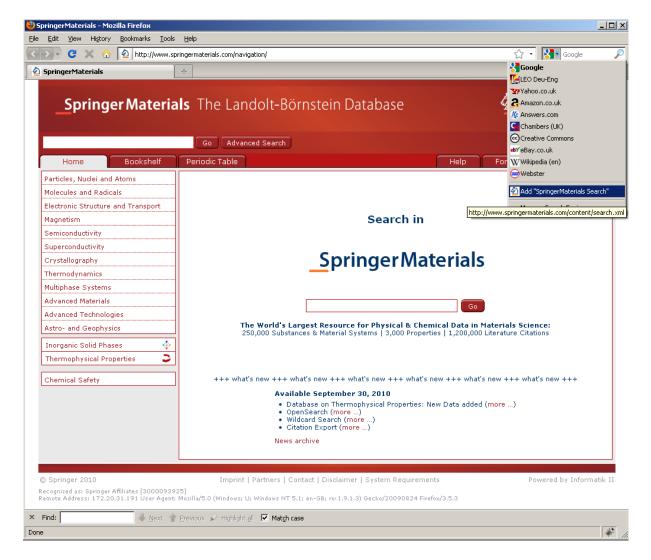
Typical Search Hits (search for methanol "vapor pressure", hits in "Thermophysical Properties" and Landolt-Börnstein data):

Springer Materia	als The Landolt-Börnstein Database 🙆 Springer
methanol "vapor pressure"	Go Advanced Search
Home Bookshelf	Periodic Table For Librarians Feedback
0 Particles, Nuclei and Atoms 0 Molecules and Radicals	Results 1 - 10 of 64 Documents         previous 1234567 next         Expanded View         Clear         Refine
0 Electronic Structure and Transport	Thermophysical Properties
0 Magnetism 0 Semiconductivity	Metadata - Substance: Methanol Metadata - Property: Yapor Pressure Fulltext: Methanol Thermophysical Data in the Dortmund Pure Component Properties (Yapor Pressure) Pure Component Properties (Dynamic
Superconductivity     Crystallography     45 Thermodynamics	Thermodynamics > Thermodynamical Properties > Organic Compounds > Vapor Pressure and Antoine Constants > Oxygen Containing Organic Compounds compounds C2C8 3 i
2 Multiphase Systems 12 Advanced Materials	Metadata - Substance: Methanol furanmethanol Oxiranmethanol Metadata - Property: vapor pressure Fulltext: Inorganic Compounds 2 Tabulated Data on Vapor Pressure of Oxygen Containing Organic Compounds
1 Advanced Technologies 0 Astro- and Geophysics	Thermodynamics > Mechanical Properties > Viscosity > Pure Organic Liquids C 🔹 i Metadata - Substance: Methanol Methanol -d Methanol -d Methanol -OD Fulltext: 300.0 n /(mPa s) 0.0497 (at saturation vapor
0 Inorganic Solid Phases 💠 1 Thermophysical Properties D	pressure) A table is given in the original 0.2123 0.1629 0.1237 (at saturation vapor pressure) A table is given in the original n/(mPa s) 0.5248 0.4207 (at saturation vapor pressure) 273.15 0.5248 trichloro-fluoro-methane
Chemical Safety	Advanced Technologies > Energy > Renewable Energy Renewable carbon-based transportation fuels 1
	Metadata - Substance: Methanol Methanol cluster CH3(OH) (methanol) Fulltext: pressure, it boils at -25°C and has a vapor pressure of 6 bar at 25°C (30 bar design solubility in hydrocarbons, and a lower vapor pressure. There are several processes which can
	Thermodynamics > Thermodynamical Properties > Organic Compounds > Enthalpies of Fusion and Transition Organic Compounds, C1C3 i i Metadata - Substance: Methanol Methanol Methanol - D1 Fulltext: ± 0.02 9.75 ± 0.13 ns
	pressure isoperibol isoperibol DSC adiabatic vapor pressure DSC DSC adiabatic vapor pressure cr   271.95 ± 0.2 22.3 ± 0.6 sa;fd vapor pressure 38-bur/slo Dichlorodimethylsilane
	Thermodynamics > Thermodynamical Properties > Organic Compounds > Vapor Pressure and Antoine Constants > Oxygen Containing Organic Compounds
	Compounds C9C57 🐚 👔 Metadata - Substance: Biphenylmethanol diphenylmethanol Metadata - Property: vapor pressure
	Advanced Materials > Dolymers > Dolymer Solutions: Drycinal Drynerties and their Delations > Thermodynamic Drynerties:

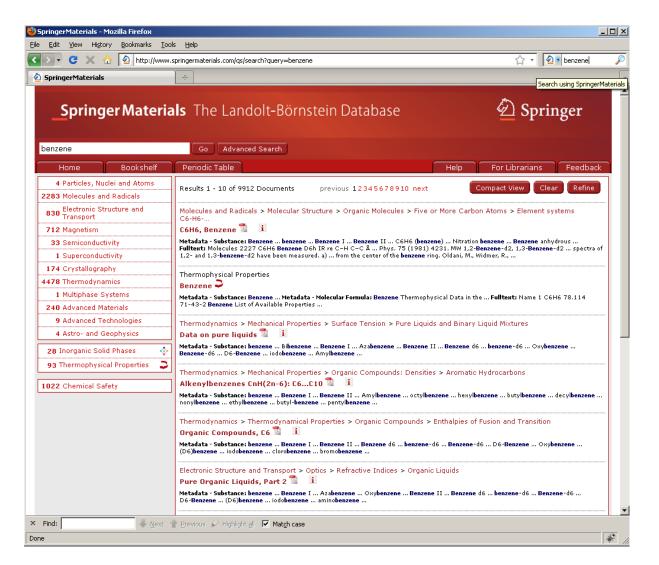
#### **Search Plugin for Web Browsers**

As an additional feature, SpringerMaterials offers a search plugin for web browsers which support OpenSearch plugins, such as Internet Explorer (as of version 7), Firefox (as of version 2) and Chrome.

With Internet Explorer and Firefox, you can add the plugin to the browser search form by selecting "Add SpringerMaterials Search" from the top-right drop-down menu. Keep in mind that this works only while you are visiting www.springermaterials.com:



After this installation, you can use the browser search form to type queries which will then be sent directly to SpringerMaterials:



If you are using another system, please check the documentation of your browser to see if it supports search plugins according to the OpenSearch 1.1 specification, and how to install them.

#### **Simple Search**

The Simple Search field is found in the center of the SpringerMaterials homepage and replicated as such in a field below the SpringerMaterials logo.

Typing effort for query formulation is reduced by suggestions of terms (Speed Typing) showing available content.

_springer Materia	als The Landolt-Bör	nstem Database			inger
	Go Advanced Search				
Home Bookshelf	Periodic Table		Help	For Librarians	Feedback
Particles, Nuclei and Atoms					
Molecules and Radicals					
Electronic Structure and Transport					
Magnetism		Search in			
Semiconductivity					
Superconductivity					
Crystallography		SpringerMat	erials		
Thermodynamics			criais		
Multiphase Systems					
Advanced Materials		nucl magn	Go	1	
Advanced Technologies		nuclear magnetic moment	<u> </u>	<b>,</b>	
Astro- and Geophysics	The World's Larg			ice on:	
Inorganic Solid Phases 🛛 💠	250,000 Substance	magnetic properties of nuclei		on:	,
Thermophysical Properties	-1	nuclear magnetic resonance data			
		nuclear magnetic relaxation rate			
Chemical Safety	+++ what's new +++ wha	nuclear magnetic resonance (NMR)		's i	new +++
		nuclear magnetic resonance method nuclear magnetic resonance signal			
		internal nucleation magnetic field			
		11B nuclear magnetic resonance data			
		<pre>revious Page 1 of 313</pre>		next >	
		1000101010		noxe P	

You can either type your query, then click "Go" or select a term from the speed-typing list and click "Go".

To edit your current query again or to add further search criteria, click "Refine".

# **Speed Typing**

Reduces typing effort for query formulation by suggesting terms and showing available content upfront. The more you type, the shorter the list of suggestions gets.

Springer Mate	🖄 Springer							
Substances, Properties,	3D Molecules E	ibliographic References	Help	Close				
Your Query					Help For Librarians Feedback			
			Go	Clear				
Search for								
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sulphur		,,,,,						
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sulphuric acid (7664					ariala			
sulphur dioxide (74	46-09-5; O2S)				erials			
sulphur mustard (50	)5-60-2; C4H8Cl2S)							
Search in sulphur nitride (289	50-34-7; N4S4)							
📃 Particle: sulphurous acid (77					Go			
Molecul <sup>®</sup> Sublimed sulphur (7	704-34-9; 9035-99-8; S)							
Electror sulphur trioxide (74	46-11-9; O3S)				cal Data in Materials Science: s   1,200,000 Literature Citations			
Semicol sulphuric diamide (3	7803-58-9; H4N2O2S)							
Superco Flowers of sulphur (		)						
Crystall Sulphuryl fluoride (;	2699-79-8; F2O2S)							
Multipha sulphur (IV) oxide (			w +++ what's new +++ what's new +++					
📃 Advanc <mark>, sulphur</mark> (VI) oxide (								
Advance sulphur dichloride (:	L0545-99-0; Cl <sub>2</sub> S)				re)			
Astro- a sulphuryl chloride (	7791-25-5; Cl <sub>2</sub> O <sub>2</sub> S)				ore) Its (more)			
🛛 🔲 Inorgan <mark>sulphuric [2H2]acid</mark>	(13813-19-9; D <sub>2</sub> O4S)							
Thermo < previous			next >					
🔲 Chemical Safety								
		it   Partners   Contact   Disc						

#### **Subject Area Navigation**

SpringerMaterials content is organized in 12 Subject Areas (see homepage):

- Particles, Nuclei and Atoms
- Molecules and Radicals
- Electronic Structure and Transport
- Magnetism
- Semiconductivity
- Superconductivity
- Crystallography
- Thermodynamics
- Multiphase Systems
- Advanced Materials
- Advanced Technologies
- Astro- and Geophysics

Click on one of the Subject Areas to move to the content level, a list of Sub-Areas will open in the main window. Red headlines offer further Sub-Sub-Areas. A "PDF" icon (1) shows that you have reached the content level; the nearby "i"-icon (1) opens the InfoPage.

_Springer Materia	🖄 Springer			
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Home Bookshelf	Periodic Table	Help For Librarians Feedback		
Particles, Nuclei and Atoms	> Magnetism			
Molecules and Radicals Electronic Structure and Transport Magnetism Semiconductivity Superconductivity Crystallography Thermodynamics Multiphase Systems Advanced Materials Advanced Materials Advanced Technologies Astro- and Geophysics Inorganic Solid Phases Thermophysical Properties Chemical Safety	Magnetism         Diamagnetic Susceptibilities         Organic Compounds: Diamagnetism and Magnetic Anisotropy         Coordination and Organometallic Compounds         Transition Metals         Elements, Mutual Alloys and Compounds         Alloys and Compounds with Main Group Elements         Rare Earth Metals         Elements, Hydrides and Mutual Alloys         Compounds with Main Group Elements         Elements, Alloys and Compounds (Update)         Actinides         Elements, and Compounds (Update)         Actinides         Elements, and Compounds, Lanthanide and Actinide Compounds         Non-Irron Garnets, Spinels and Hexagonal Ferrites         Index of substances         Garnets and Perovskites (Update)         Fe Oxides and Compounds, Spinels (Update)         Hexagonal Ferrites, Lanthanide and Actinide Compounds (Update)         Non-Irron Garnets, Spinels and Hexagonal Ferrites         Index of substances         Garnets and Perovskites (Update)         Fe Oxides and Compounds, Spinels (Update)         Hexagonal Ferrites, Lanthanide and Actinide Compounds (Update)         Non-Metals	s, Perovskites Compounds, Lanthanide and Actinide Compounds, Perovs		
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#### Thermophysical Properties (DDBST – Dortmund Data Bank Software & Separation Technology Database)

The subset of the DDBST (Dortmund Data Bank Software & Separation Technology) contains thermophysical properties of the 50 most important organic liquids plus water and their 1225 binary mixtures.

The database can be browsed and is accessible through all search functions. A logo ( $\checkmark$ ) next to a Search Hit indicates the source.

_Sprin	ger Materi	<b>als</b> The Lan	dolt-Börnstein Database		🖄 Sprin	nger
		Go Advan	ced Search			
Home	Bookshelf	Periodic Table		Help	For Librarians	Feedback
Particles, Nuclei Molecules and R. Electronic Struct Magnetism Semiconductivit, Superconductivit Crystallography Thermodynamic: Multiphase Syste Advanced Materi Advanced Techn Astro- and Geop Inorganic Solid I Thermophysical Chemical Safety	adicals ure and Transport y y s ms als ologies hysics 2hases		Search in Thermophysical Pu Substances / Molecular Formulas / CAS [ Substances & Water   1,225 Binary Mixtures   350,0 About Thermophysical Pro	S Registry Num	bers Go	ations
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Typical query for Thermophysical Properties:

Go Advanced Search								
Home Bookshelf	Periodic Table	Hel	p For Librarians Feedba					
Particles, Nuclei and Atoms								
Molecules and Radicals								
Electronic Structure and Transport								
Magnetism		Search in						
Semiconductivity								
Superconductivity		The survey have been been been as						
Crystallography		Thermophysical Proper	Thermophysical Properties					
Thermodynamics								
Multiphase Systems								
Advanced Materials		Substances / Molecular Formulas / CAS Registry						
Advanced Technologies		benzene	Go					
Astro- and Geophysics	50 Organic Su	Benzene (71-43-2; C6H6)	00 Literature Citations					
Inorganic Solid Phases 🛛 💠	il	Benzene (71-43-2; C6H6) + Water (7732-18-5; H2O) Acetone (67-64-1; C3H6O) + Benzene (71-43-2; C6H6)						
Thermophysical Properties		Benzene (71-43-2; C6H6) + Hexane (110-54-3; C6H14)						
mermophysical Propercies	1	Benzene (71-43-2; C6H6) + Methanol (67-56-1; CH4O)						
Chemical Safety		Benzene (71-43-2; C6H6) + Phenol (108-95-2; C6H6O)						
		Benzene (71-43-2; C6H6) + Toluene (108-88-3; C7H8)						
		Ethanol (64-17-5; C2H6O) + Benzene (71-43-2; C6H6)						
		Benzene (71-43-2; C6H6) + Decane (124-18-5; C10H22)						
		Benzene (71-43-2; C6H6) + Pentane (109-66-0; C5H12)						
		< previous Page 1 of 5	next >					

Typical overview for Thermophysical Properties:

-	Sprir	nger Ma	aterials The l	andc	lt-Börnstein Database 🙆 Springer
The	izene / M rmophysic ponents		he Dortmund Data Ban	k	
		Molar Mass	CAS Registry Number	Name	
1	C6H6	78.114	71-43-2	Benzene	
2	CH4O	32.042	67-56-1	Methanol	
Solic Exce Dens	ure Viscosit I-Liquid Equ ss Volumes sities of pr-Liquid Eq Dortr	ilibria Ə ; Ə uilibria Ə	nk Edition - 2010. Integra	ted in Spr	ngerMaterials - 2010
	Dortr © Sp	nund Data Ba	nk by DDBST - Dortmund ST GmbH 2011. All Rights	Data Ban Reserved	Software and Separation Technology GmbH Version 2010.10.
	inger 2010			print   Par	ners   Contact   Disclaimer   System Requirements Powered by Informatik
			3000093925] Jser Agent: Mozilla/5.0 (Wind	ows; U; Wir	dows NT 5.1; en-GB; rv:1.9.1.3) Gecko/20090824 Firefox/3.5.3

# **Your Query**

A field in the Advanced Search that combines all search strings typed in any of the other fields of the Advanced Search page into a Boolean query that you can either submit as is or adapt to your needs before submitting to the search engine.

_Springer Materi	als The I	_andolt-Börnstein	Database	🖄 Springer
Substances, Properties,	3D Molecules	Bibliographic References	Help Close	
Your Query				Help For Librarians Feedback
{"sulphur dioxide" or "7446-09-5" or	r "02S"} {"virial c	oefficient"}	Go Clear	Expanded View Clear Refine
Search for				
Substances / Molecular F "sulphur dioxide" or "74		nt Systems / CAS Registry Numb	ers	3-2 67-56-1 Metadata - Molecular Formula: C6H6
Properties	40-09-5 Ur UZS			Registry Number Name 1 C6H6 78.114 71-43-2 Benzene
"virial coefficient"				
virial coefficient				
third virial coefficient				
Search in second virial coefficient				
Particle: < previous				
Molecules and Radicals Electronic Structure and Transpo	.+			
Magnetism		ore of these words		
Semiconductivity				
Superconductivity	exactly t	his phrase		
Thermodynamics				
Multiphase Systems	but none	of these words		
Advanced Materials				
Astro- and Geophysics				
🔲 Inorganic Solid Phases				
Thermophysical Properties				
🔲 Chemical Safety				s Powered by Informatik II
Remote Address: 172.20.31.191 User Agen	nt: Mozilla/5.0 (Wind	lows: U: Windows NT 5.1: en-GB: rv: )	.9.1.3) Gecko/20090824 Firef	ox/3.5.3

[up]

# Wildcards

SpringerMaterials supports single and multiple character wildcard searches within terms. Use the "?" symbol to perform a single character wildcard search and use the "\*" symbol to perform a multiple character wildcard search.

For example, the query *sul\*ite* returns results for both "sul*ph*ite" and "sul*f*ite". Searching for *ferr?magnetism* shows hits for both "ferr*o*magnetism" and "ferr*i*magnetism".